



# Laser-induced graphenization of poly(dimethylsiloxane)/poly(ethylene glycol) composite

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## INTRODUCTION

With the accelerating development of wearable smart electronics, flexible strain sensors are in rising demand. Cost-effective, single-step preparation of laser-induced graphene (LIG) on polymeric surfaces has emerged as a highly promising technology for fabricating wearable LIG-based sensors [1]. The reason for that are remarkable properties of LIG such as good flexibility, piezoresistive properties, high mechanical stability, good electrical conductivity, low cost and ease of preparation [2]. However, the single-step fabrication of LIG/polymer composites that are both biocompatible and conductive, as well as stretchable, remains a significant challenge [3]. Poly(dimethylsiloxane) represents biocompatible and stretchable polymer being the most popular elastomer because of its great mechanical properties [4]. Because of its lack of aromatic structure and low carbon content, pure PDMS is not suitable for LIG so our study addresses preparation of laser induced graphene on a novel biocompatible poly(dimethylsiloxane)/poly(ethylene glycol) (PDMS/PEG) materials using optimal processing conditions.

## RESULTS

Illustration of laser induction on PDMS/PEG composite is shown in Figure 1. SEM image of the surface of LIG on PDMS with 40 wt.% of PEG is presented on Figure 2. Raman spectra of LIG on PDMS/40%PEG is presented in Figure 3. Graph  $I_D/I_G$  and  $I_{2D}/I_G$  as a function of percentage of PEG in PDMS is presented on Figure 4. FTIR spectra of LIG on PDMS with 10 and 40 wt.% are presented on Figure 5 and band assignation in Table 1.



Figure 1. Illustration of laser induction on PDMS/PEG

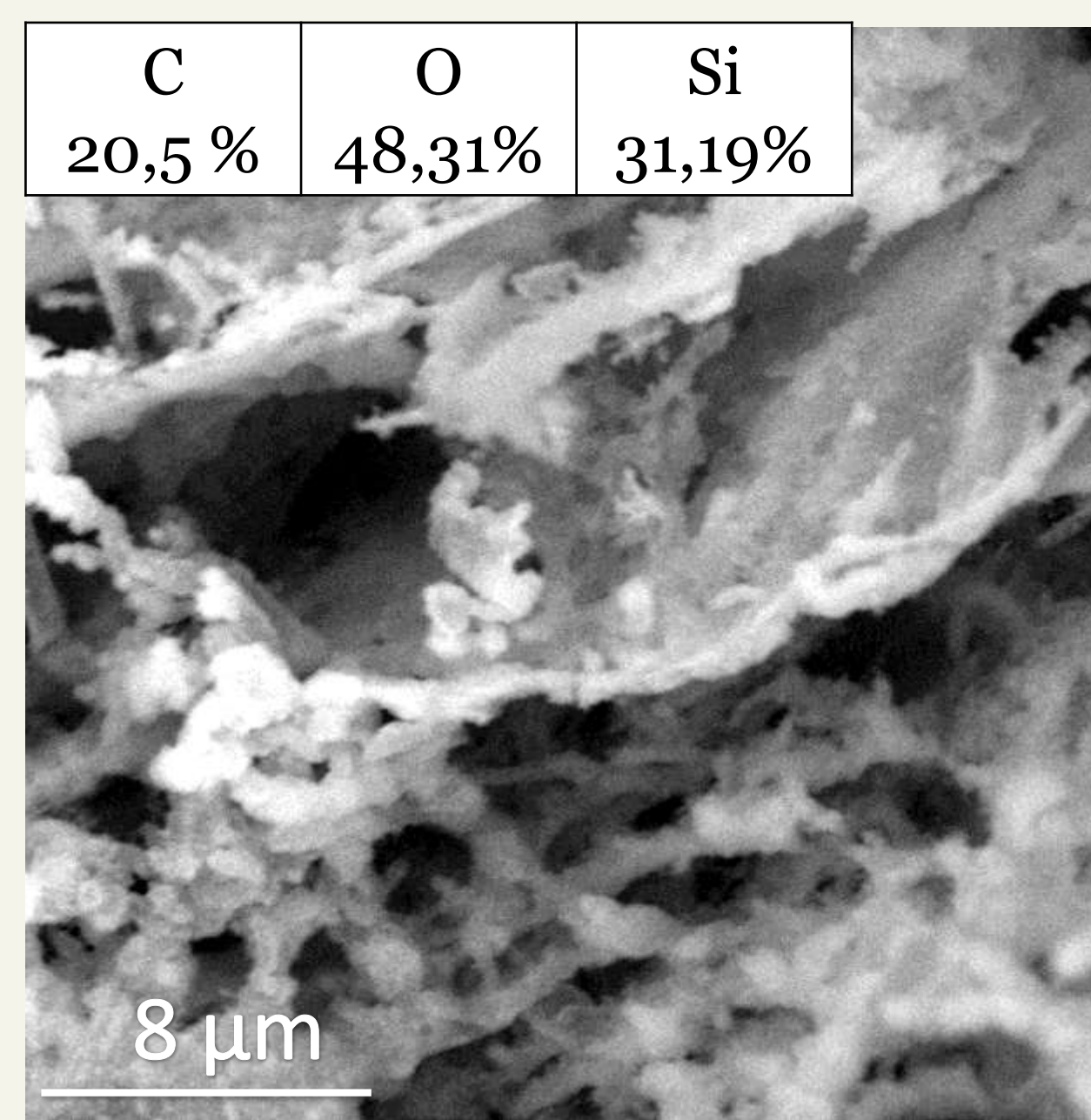


Figure 2. SEM micrograph with EDX analysis of LIG on PDMS with 40 wt.% PEG with 10 000x magnification

## DISCUSSION

The results revealed that adding PEG to PDMS is favorable for induction of graphene. Methods of characterization have confirmed the formation of graphene on the surface of PDMS/PEG. SEM micrographs (Figure 2) showed expected 3D porous structure of LIG. FTIR and SEM/EDX analyses also indicated the presence of SiO<sub>2</sub> nanoparticles, attributed to the thermal degradation of the PDMS matrix under CO<sub>2</sub> laser treatment. The study of Raman spectra (Figure 3) show three prominent peaks that are typical features of graphene: strong G band shown at ~1585 cm<sup>-1</sup> originating from a first order zone-boundary phonons and presence of sp<sub>2</sub> carbon bonds, D band at ~1348 cm<sup>-1</sup> induced by defects and vacancies and 2D band at ~2700 cm<sup>-1</sup> that originates from second order zone-boundary phonons [2]. The analysis of the intensity ratio of the D and G peaks and 2D and G peaks was performed (Figure 4) and the lowest intensity ratio was obtained for LIG on PDMS with 40 wt.% of PEG indicating bigger crystal domains in graphene. Two more measurements of  $I_D/I_G$  and  $I_{2D}/I_G$  were picked to represent the instability of the sample with 50 wt.% of PEG in PDMS.

## METHODS AND MATERIALS

Poly(dimethylsiloxane) (Dow Corning, Sylgard 184) and poly(ethylene glycol) (Sigma Aldrich, Mn=400 g/mol) were used. Five sets of PDMS samples were synthesized with different weight concentrations of PEG (10-50 wt.%). The CO<sub>2</sub> laser used to produce LIG was a DBK FL-350. FTIR spectra of PDMS/PEG and LIG were recorded with a Thermo Fisher Scientific, Waltham, MA, USA spectrometer, in transmission mode on a KBr substrate. Raman spectra of the samples were recorded with a DXR Raman microscope (Thermo Fisher Scientific, Waltham, MA, USA). SEM imaging with EDX was done at magnifications 400x, 1000x, 2000x, 5000x and 10000x with a PhenomProX scanning electron microscope (Phenom, Thermo Fisher Scientific, Waltham, MA, USA).

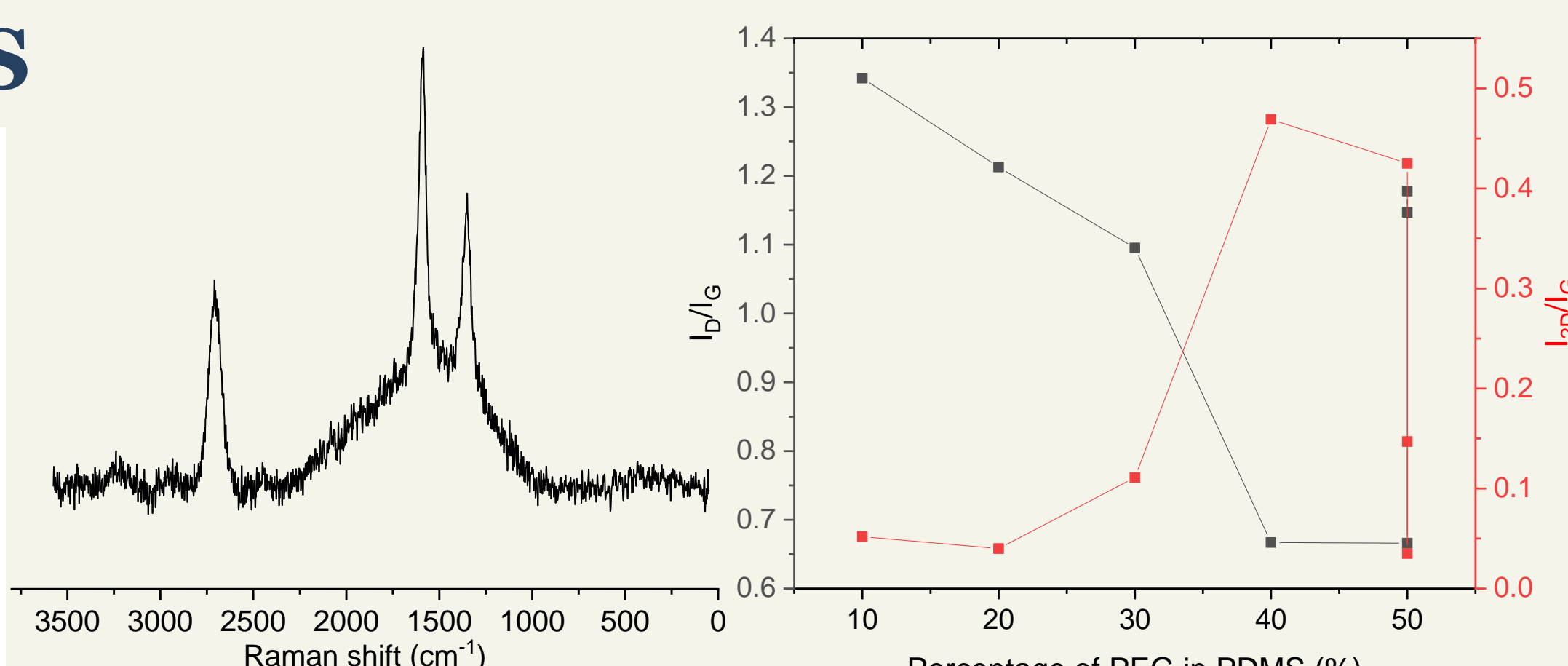


Figure 3. Raman spectrum of LIG on PDMS with 40 wt.% PEG

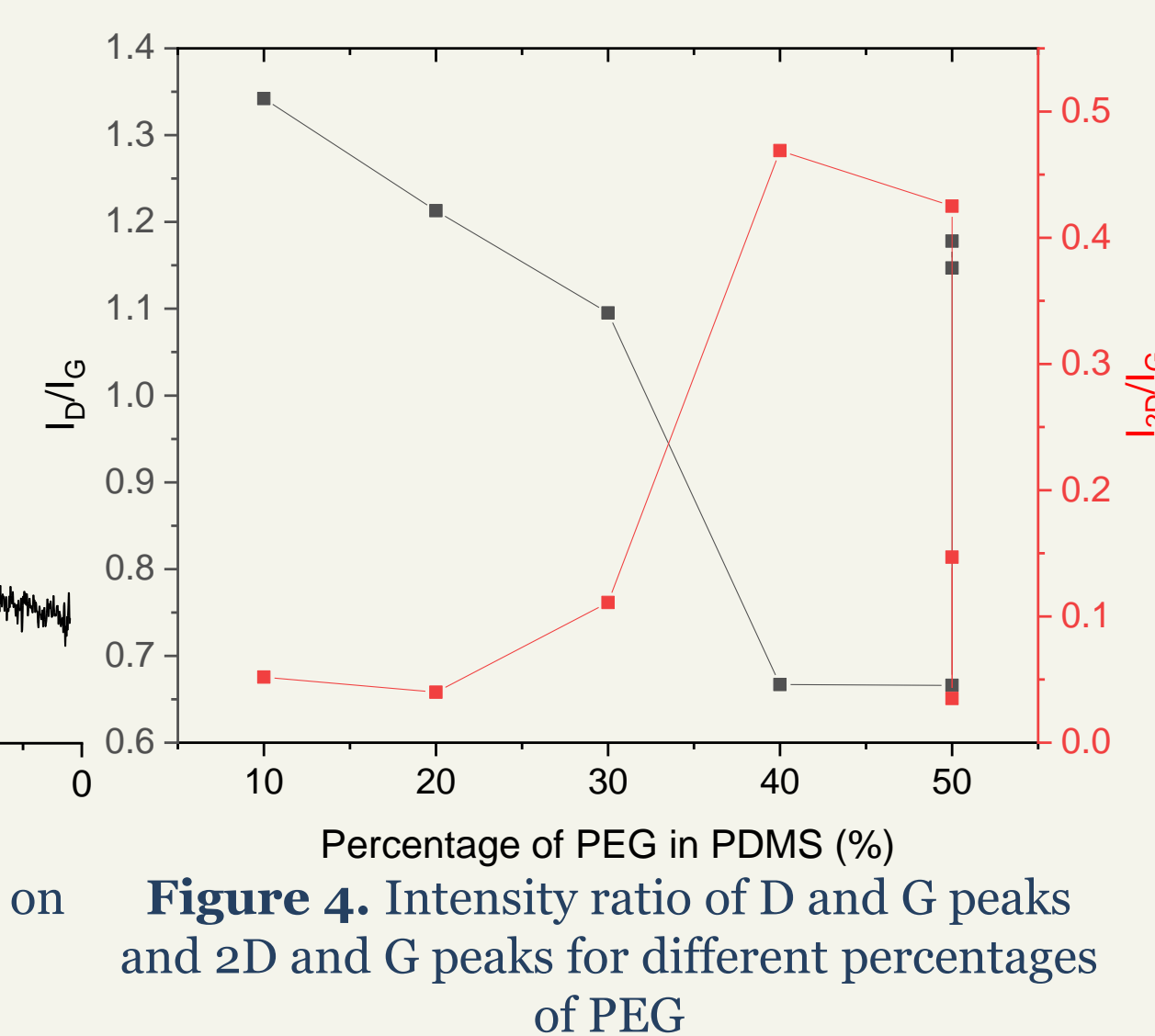


Figure 4. Intensity ratio of D and G peaks and 2D and G peaks for different percentages of PEG

## CONCLUSIONS

- Graphene can be induced by laser irradiation of PDMS/PEG composites
- The resultant graphene is characterized with FTIR, SEM/EDX and Raman spectroscopy
- 40 wt.% PEG concentration is found to be optimal for induction of graphene, similar to what was observed earlier with PEEK/PDMS composites [5]
- The materials also contains SiO<sub>2</sub> and SiC similar to what was observed earlier with PEEK/PDMS composites [5].

## REFERENCES

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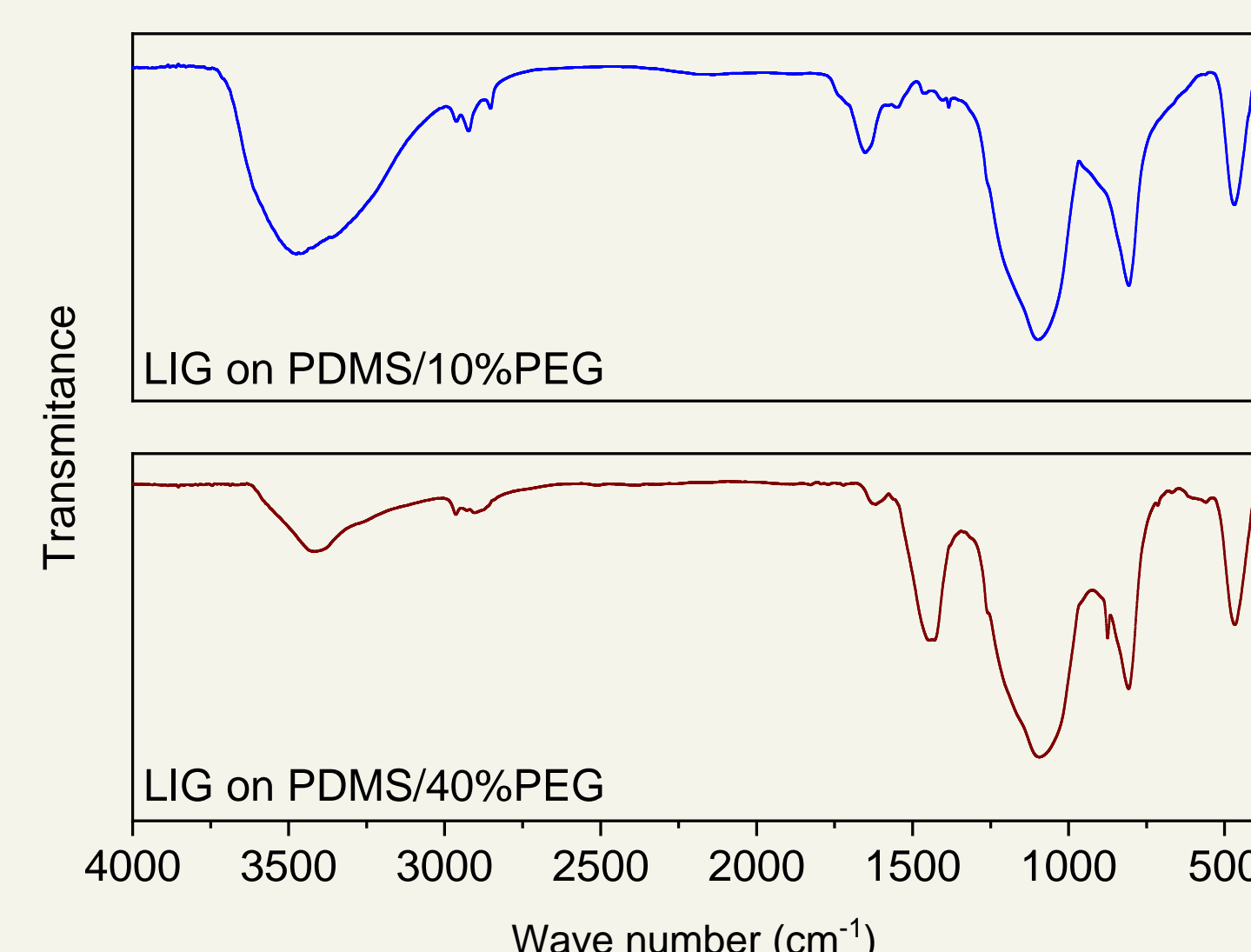


Figure 5. FTIR spectra of LIG on PDMS/10%PEG and LIG on PDMS/40%PEG

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Table 1. Band assignment of LIG on PDMS with 10 wt.% PEG and LIG on PDMS with 40 wt.% PEG

	$\nu(\text{O-H})$ (cm <sup>-1</sup> )	$\nu(\text{CH})_{\text{as}}$ (cm <sup>-1</sup> )	$\nu(\text{CH})_s$ (cm <sup>-1</sup> )	$\nu(\text{C=C})$ (cm <sup>-1</sup> )	$\delta(\text{C-H})_s$ (cm <sup>-1</sup> )	SiO <sub>2</sub> nanoparticles + $\nu(\text{C-O})$ (cm <sup>-1</sup> )	Si-C ili C-H rocking (cm <sup>-1</sup> )	$\delta(\text{C-C})$ (cm <sup>-1</sup> )
PDMS/ 10%PEG	3331	2962	2866	1629	/	1075	801	/
PDMS/ 40%PEG	3313	2962	/	1618	1259	1073	800	461

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